

FIELD-WIDE APPLICATION OF CHEMICAL TRACERS FOR MASS FLOW MEASUREMENTS IN PHILIPPINE GEOTHERMAL FIELDS

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Keywords : Chemical Tracers, MgCl₂, Na benzoate, Na fluorescein, SF₆

Abstract

Chemical tracers are widely utilized in Philippine geothermal fields to accurately measure the bore output of production wells, injection line flow rates, and steam inlet rates to power plants. Based on these flow measurements, a production-injection well utilization strategy is formulated to optimize production and minimize injection fluid breakthrough to the production sector. The reservoir response to field utilization is fully assessed based on the total steam and brine flow measurements with time and individual injection well capacity trends.

Recent on-line bore output measurements using sodium benzoate for brine flow and sulfur hexafluoride for steam flow demonstrated a good correlation with the James tube method for several production wells tested. Similarly, steam flow rates measured by tracer compared well with venturi flowmeter measurements at power plant inlets.

Magnesium chloride has been extensively used as a chemical tracer for water flow measurements by PNOC. These test results yielded close agreement with orifice plate flowmeter and spinner flow measurements. Other chemical tracers, such as sodium benzoate and sodium fluorescein, were compared with the magnesium chloride tracer and produced similar results.

1.0 INTRODUCTION

The chemical tracers used in PNOC-EDC geothermal fields in the Philippines include magnesium chloride (MgCl₂), sodium fluorescein (Na-fluorescein), and sodium benzoate (Na-benzoate) for brine flow measurements and sulfur hexafluoride (SF₆) for steam flow. Of the tracers mentioned, MgCl₂ has been extensively applied since the early 1990's because of its proven good correlation with flowmeter, orifice plate, and downhole Spinner methods. In 1997, PNOC-EDC started to apply the simultaneous injection of Na-benzoate and SF₆ in two-phase lines. Na-benzoate produced results similar to MgCl₂ tests, while steam flow measurement by SF₆ correlated with venturi meter. On the other hand, the application of Na-fluorescein is still not implemented due to its relative lower precision and higher costs.

The objective of this paper is to determine the accuracy and the precision of the above-mentioned chemical tracers. They are compared with standard methods, like James Lip Pressure for two-phase tests, venturi meter for the steam flow, and flowmeter for the brine flow. It is also the objective of this paper to identify possible errors for each method and quantify them through error bands.

2.0 METHODOLOGY

2.1 Field Testing

Chemical tracers are injected into the pipelines. A distance of 40 times the pipe diameter is maintained between the injection point and sampling point to attain homogeneous mixing of the tracer and the fluids. For $MgCl_2$ and Na-fluorescein, tracers are mixed in a 20 liter pressure vessel, which are injected into the pipe through the help of pressurized gas. For Na-benzoate and SF_6 , an electronically controlled equipment is utilized to inject the tracers into two-phase lines. This is called in this paper as TFT or Tracer Flow Test. Sufficient number of samples are collected to establish breakthrough curves. For $MgCl_2$, it is customary to collect 20 samples for a period of 6.6 minutes. For Na-benzoate, it is a standard procedure to collect 5 samples, and for SF_6 4 gas samples are collected per trial run.

2.2 Laboratory Analyses of Chemical Tracers

Na-benzoate is analyzed using High Performance Liquid Chromatography (HPLC) with a W detector set at 230 nm and methanol-phosphate buffer as eluent. Concentrations of as low as 0.1 ppm can be analyzed although the usual working range is from 10.0 ppm to 100.0 ppm. Magnesium is determined using atomic absorption spectrophotometer (AAS) at 285.2 nm wavelength with nitrous oxide-acetylene flame. Sample concentrations range from 2.0 ppm-5.0 ppm which is at least ten times the background levels found at LGPP wells. Fluorescein analysis is done using a spectrofluorophotometer set at an excitation wavelength of 450 nm and emission wavelength of 512 nm. Concentrations in parts per billion (ppb) levels can still be detected although the usual sample concentration range is 0.1 ppm to 2.0 ppm.

SF_6 gas is analyzed in HP 6890 gas chromatograph with electron-capture detector (GC/ECD) and argon-methane mixture as carrier gas. SF_6 concentrations of as low as $6.00E-6$ ppm can still be detected although usual sample concentration range is from 0.001 to 0.100 ppm. Quantitative determination of tracer samples is made using a spreadsheet developed by Thermochem, Inc.

3.0 DISCUSSION OF RESULTS

The following sections compare the results of the tracer tests with the accepted field methods. Likewise, the tracer methods are compared with each other to determine their advantages and disadvantages. Tracer tests were conducted in Leyte Geothermal Power Project (LGPP), a geothermal field operated by PNOC-EDC situated in Central Philippines.

3.1 TFT: Two-Phase Flow Measurement

3.1.1 Steamflow : TFT vs. Venturi Meters

Shown in Table 1 are the results of the comparative testing between TFT and Venturi meter. The testing was conducted in a separator vessel at Malitbog Sector of LGPP. The SF_6 was injected at the two-phase line leading to the separator vessel and gas samples were collected at the steam header. Simultaneously, for every gas sample taken, a venturi meter reading was taken. As shown in the table, the maximum percentage difference between the two is $\pm 4\%$.

Separator Vessel	Date	Average Steamflow by TFT, (kg/s)	Average Steamflow by Venturi, (kg/s)	Percent Difference, (%)
SV-501	19-Jul-97	64.63	62.07	4.04
	22-Jul-97	79.43	76.2	4.15
	24-Oct-97	82.1	80.32	2.19
	31-Oct-97	65.58	68.4	-4.21

Table 1: Comparison of steamflow measurement by TFT and Venturi Meter.

3.12 Two-Phase Flow: TFT vs. James Lip Pressure Method

Table 2 shows the **results** of bore output testing by James Lip Pressure Method and TFT done simultaneously on the same well. There is close correlation between the absolute values of results obtained. The **difference** between output parameters are as follows: around 3 kg/s for mass flow; around 60 kJ/kg for enthalpy; and almost 1 kg/s for steamflow. Unfortunately the wells *tested* have **minimal** waterflow, that is why, a **difference** of about 4 kg/s was obtained for the water flow.

Production Well	Method	WHP, (Mpa)	Enthalpy, (kJ/kg)	Steam Flow, (kg/s)	Water Flow, (kg/s)
213	TFT	0.97	2001	19.6	11.3
	JLPM	1	2191	19.6	7.5
212	TFT	1.3	2712	35.1	1.1
	JLPM	1.1	2678	34	0*

*Weirbox level very low (not measurable by JLPM)

Table 2: Bore output measurement by TFT and James Lip Pressure Method

3.2 MgCl₂ : Brine-flow Measurement

3.2.1 MgCl₂ vs Turbine Flowmeter

The use of MgCl₂ tracer started as an alternative method to the **orifice** plate and flowmeter in the early 1990's, (Macambac, 1992). Precision and accuracy were **tested** for 1.3 years (August 1991 to December 1992) parallel with flowmeter. Table 3 shows that the capacity test results of the flowmeter and MgCl₂ tracer have close agreement falling within **5%** difference. **Because** of the **good** correlation, thenceforth, the MgCl₂ tracer **has** been used in all brine flow measurement in reinjection wells and load in main lines.

Production Well	DATE	MgCl ₂ Tracer, (kg/s)	MWD Flowmeter, (kg/s)	Percent Difference (%)
1R3	Sep-91	39.9	41.4	3.7
	Jun-92	59.6	59.3	0.5
	Sep-91	61.1	57.9	5.4
1R4DA	Jun-92	87.1	82.3	5.7
	Dec-92	75.7	76.6	1.2
1R5D	Oct-91	33.6	33.4	0.6
	Jun-92	57.2	59.7	4.3
	Sep-92	64.8	64.1	1.1
1R8D	Oct-91	125.7	124.4	1.0
	Mar-92	116	119	2.6
	Juni-92	120.2	116.8	2.9
2R4D	Oct-91	18.6	17.9	3.8
	Jun-92	16.2	16.3	0.6

Table 3: Calibration of MgCl₂ tracer with the flowmeter

3.2.2 MgCl₂ vs Spinner Survey

Selected waterflow measurements were **conducted** in LGPP wells simultaneously with MgCl₂ tracer and Spinner survey. Although only few **tests** were **conducted**, the **results** at fluid flow rate of less than 30 kg/s show that spinner **surveys** and MgCl₂ **are** comparable within a flow difference of less than < 5.0 kg/s (Table 4). Note also that **all** spinner **survey results** are below the **MgCl₂ results**.

3.23 MgCl₂ vs Na-fluorescein Tracer

Disodium fluorescein dye (C₂₀H₁₀Na₂O₅), has been routinely mixed in small amount with MgCl₂ tracer solution to determine whether the injected brine has already reached the sampling location. This can be determined by directly observing the changes of color of the brine from colorless to yellowish signaling the start of sampling. Since fluorescein dye is always used during every MgCl₂ tracer testings, a study was conducted to determine its ability to quantitative waterflow measurement (Paraon, 1997).

Production Well	WHP (MPa)	Spinner Survey, (kg/s)	MgCl ₂ Tracer (kg/s)	Difference (kg/s)
5R3D	1.50	23.3	23.4	0.01
5R5D	1.40	21.1	25.8	4.7
5R8D	1.50	20.4	24.9	4.5

Table 4: Comparison of MgCl₂ and Spinner Survey Results (May 1994)

Comparison of MgCl₂ tracer with fluorescein dye show close agreement for 70 percent of the tests conducted based on the 5 % relative difference limit criterion (Table 5). In other testings of the fluorescein dye, the tracer falls short in terms of precision with the MgCl₂. The reason could be due to the irreversible thermal decay at temperature greater than 260 °C, or to the dye's sensitivity to ultra-violet rays.

Reinjection Well/Line	WHP MPa	Average WaterFlow, (kg/s)		Percent Relative Difference (%)
		MgCl ₂ Tracer	Na-Fluorescein Tracer	
5R7D	1.08	49.6	51.9	4.53
	1.04	49.0	49.8	1.62
MG-9RD	0.23	139.4	138.3	0.79
MahMRL	0.52	94.3	92.4	2.04
MG-21D	0.75	108.0	107.3	0.65
MN-3RD	1.80	23.5	20.2	15.1
Sam MRL	0.52	68.4	55.0	21.7

Table 5: Comparison of waterflow measurements by MgCl₂ and Na-fluorescein tracers

3.24 MgCl₂ vs Na-benzoate

Table 6 summarizes the results of waterflow based on the simultaneous testings of MgCl₂ and Na-benzoate tracers conducted in several reinjection wells and main reinjection line. The accuracy of Na-benzoate over the MgCl₂ falls within the acceptable relative difference of 5%.

4.0 VARIANCES AND ERRORS

4.1 Error Bands for MgCl₂

Variance for MgCl₂ flow measurements at different reinjection loads were also established, taking into consideration both the field and laboratory errors (Siega and Arones, 1996). Figure 1 shows that for higher flows rates between 140-225 kg/s, the variance is between ±10-15 kg/s, while at lower flows between 25-100 kg/s, the variance is between ±5-7 kg/s. Shown also in the graph is the general equation of the regression line.

Reinjection Well/Line	Waterflow, (kg/s)		% Rel Difference
	MgCl ₂ Tracer	Na-benzoate Tracer	
Sam MRIL	83.5	82.2	1.56
	96.9	96.0	0.93
MG-7RD	177.1	182.2	2.88
5R7D	47.1	47.4	0.64
4R3D	97.0	100.3	3.40
5R4D	177.8	170.1	4.33
5R1	79.2	74.9	5.43
5R5D	22.8	23.8	4.39
5R3D	23.9	24.8	3.77

Table 6: Comparison of waterflow measurements by MgCl₂ and Na-benzoate tracers

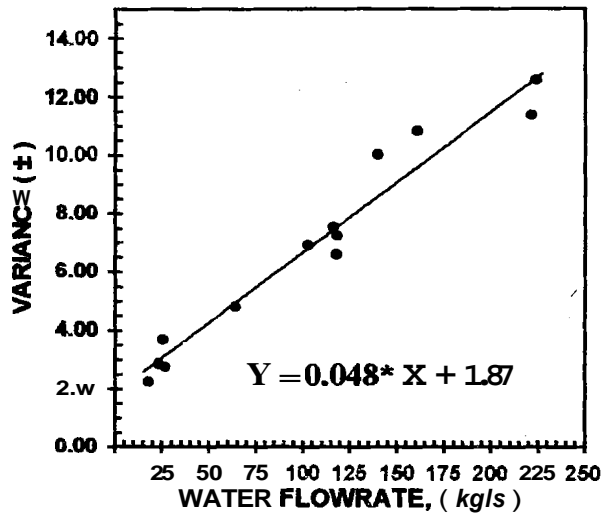


Figure 1: Variance in brineflowrates in MgCl₂ tracer

4.2 Error Bands for SF₆ and Na-benzoate

Tabulated in Table 7 are the estimated maximum errors for water flow, steam flow and enthalpy associated with TFT testing using sodium benzoate and SF₆ tracers

Measurement	Error	Max Percentage error (%)
Steamflow	Gas Tracer Analytical	2.0
	Gas Tracer Injection Rate	2.0
	Mixed Sample Analytical	2.0
Cumulative		3.5
Waterflow	Liquid Tracer Analytical	1.0
	Liquid Tracer Injection Rate	2.0
	Mixed Sample Analytical	1.0
Cumulative		2.4
Enthalpy		1.7

Error for the water flow measurement is **2.4%** while error for the steam flow is **3.5%** for flow rates ranging from **20 kg/s** to **300 kg/s**. These **values** are the possible maximum errors under normal testing **conditions**. These errors contribute to maximum of 1.7% error for **calculated** enthalpy for **values** ranging from **700** to **2800 kJ/kg** (Villa and Magdadaro, 1997).

5.0 FIELD APPLICATION OF CHEMICAL TRACERS

Intensive field testing proved **that** the **results** obtained from chemical **tracer** method are comparable to those **obtained** from the **accepted** field methods. Moreover, it **has** proven that the **results** of the chemical methods **are** repeatable at the same testing condition. It is also interesting to note that a simple set-up **utilized** in MgCl_2 tracer method could give comparable **results** to a more sophisticated and electronically controlled TFT as far as brine flow measurement is concerned.

Having proven their precision and **accuracy**, chemical **tracers** **are** being extensively applied to the management of Philippine geothermal fields. The following **are** some examples of their practical applications:

5.1 Mass **flow** and **enthalpy** measurement in production wells using **TFT**

New environmental **laws** prohibit **direct** disposal of waste brine **to** the environment, and thus the James Lip Pressure method will have limited applications. In line with **this** method, TFT **has** been **adopted** to update well **bore** output for at least **twice** a year. With the updated data, it is easier to formulate production well utilization **based** on the decline or improvement of well's **mass** flow and enthalpy.

5.2 Waste brine flow measurement using MgCl_2

Inventory of brine flow to reinjection wells is imperative for a sound **reservoir** management, **because** **only** through precise and accurate measurement **can** a reinjection strategy be formulated. MgCl_2 tracer method offers a **quick** and efficient way of assessing brine load, and **has** been applied in Tongonan and Mahanagdong geothermal fields. Moreover, assessing the reinjection **capacity** of each well gives information on the decline or **gain** in injection **capacity** of the field, leading to **decision** either to drill a new well or perform work-over **operations**.

5.3 Brine audit for optimization of power plants

Bottoming cycle power plant was **installed** in Malitbog sector of LGPP, which utilized brine from main flash plants to generate an additional **14 MWe**. Prior to the plant commissioning, brine audit was **performed** using Na-benzoate and MgCl_2 tracer tests to determine the sustainability of the power plant.

In the **past** years, chemical tracer method **has** been a valuable tool in managing geothermal field **because** of their proven reliability. However, **as** in **any** tool, **this** method **must** be **constantly** improved and refined through **trials** of other chemical tracer **and** design modification.

ACKNOWLEDGMENT

The authors wish to express their appreciation to PNOC-Energy Development Corporation and Thermochem Inc. for permission to **publish** **this** paper. We also wish to thank Noel **Salonga** and Peter Barnett for the extended **support** and review of **this** project., And to the One above who guides us everything for the **success** of **this** paper.

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